

Question 1: Symmetric (Boson) and Anti-symmetric (Fermions) Wavefunction

A) Consider a system of **two fermions**. Which of the following wavefunctions can

describe the system? $\psi_{n_1, n_2}(\vec{r}_1, \vec{r}_2) = \frac{1}{\sqrt{2}} \{ \psi_{n_1}(\vec{r}_1) \psi_{n_2}(\vec{r}_2) + \psi_{n_1}(\vec{r}_2) \psi_{n_2}(\vec{r}_1) \}$ or

$\psi_{n_1, n_2}(\vec{r}_1, \vec{r}_2) = \frac{1}{\sqrt{2}} \{ \psi_{n_1}(\vec{r}_1) \psi_{n_2}(\vec{r}_2) - \psi_{n_1}(\vec{r}_2) \psi_{n_2}(\vec{r}_1) \}$. Justify your answer.

B) **Two identical particles in a box** (in one dimension 1D) is described by the **total**

wavefunction, $\psi(x_1, x_2) = \frac{1}{\sqrt{2}} \left\{ \frac{2}{L} \sin\left(\frac{2\pi x_1}{L}\right) \sin\left(\frac{4\pi x_2}{L}\right) - \frac{2}{L} \sin\left(\frac{4\pi x_1}{L}\right) \sin\left(\frac{2\pi x_2}{L}\right) \right\}$,

where L is the length of the box and x_1 and x_2 are the positions of particle 1 and 2, respectively. What are the **quantum states** of the two particles? Are the two particles **bosons** or **fermions**? **Explain your answer**. What is the **total energy** of the two identical particles? The mass of a particle is $m = 1.67 \times 10^{-27} \text{ kg}$, and $L = 1 \text{ nm}$.

Hint: A particle in a box of length, L, is described by the wavefunction

$\psi_n = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$, where $n = 1, 2, 3, \dots$ is the **quantum state** of the particle. For a

particle in the n state, its energy is $E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$.

C) Repeat part D) for the wavefunction,

$\psi(x_1, x_2) = \frac{1}{\sqrt{2}} \left\{ \frac{2}{L} \sin\left(\frac{\pi x_1}{L}\right) \sin\left(\frac{3\pi x_2}{L}\right) + \frac{2}{L} \sin\left(\frac{3\pi x_1}{L}\right) \sin\left(\frac{\pi x_2}{L}\right) \right\}$.

D) repeat for the wavefunction $\psi(x_1, x_2) = \frac{2}{L} \sin\left(\frac{\pi x_1}{L}\right) \sin\left(\frac{\pi x_2}{L}\right)$.

E) Consider **two electrons** in a box of width $L = 1.50 \text{ nm}$. The quantized energy of

each electron is $E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$, $n = 1, 2, 3, \dots$, where $m = 9.1 \times 10^{-31} \text{ kg}$ is the mass of

the electron. Calculate the **ground state energy** (lowest energy) and **first excited energy** (second lowest energy) of the **two electrons** system. Express your answer in eV ($1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$).

Question 2 Magic Numbers and Hassium

As discussed in the textbook, certain values of protons, Z and neutrons N are called "magic" since they are believed to correspond to nuclear stability. These **magic numbers** are 2, 8, 20, 28, 50, 82, and 126. However, there are other magic numbers associated only with protons or neutrons. For example, Hassium, ${}_{108}^{270}\text{Hs}$ is one of the

heaviest isotopes ever observed, and is stable for about 10s. Its surprising stability is attributed by some to be due to the fact that 108 is a magic number for protons, and 162 is a magic number for neutron.

A) Despite this explain in no more than **two sentences**, the physical reasons why ${}_{108}^{270}\text{Hs}$ should **not be stable**.

B) Use the nuclear liquid drop model (see equation sheet) to calculate the binding energy of ${}_{28}^{48}\text{Ni}$. Does the results support the magic number hypothesis? Justify in **two sentences**.

C) ${}_{108}^{270}\text{Hs}$ has atomic mass of 270.13429u. Use this to calculate the binding energy, and compare (no more than **two sentences**) with the results of part B).

Question 3 β^+ and α decay, and electron capture.

A) Consider the nuclear reaction, ${}^8_{14}\text{O} \rightarrow {}^7_{14}\text{N} + \beta^+ + \nu_e$. Calculate the **disintegration energy** of the decay using $Q = [M({}^8_{14}\text{O}) - M({}^7_{14}\text{N}) - 2m_e]c^2$. In **no more than two sentences**, explain why it is necessary to included the mass of two electrons in the equation for Q. In **one sentence**, justify, based on your answer, why the decay will or will not occur? What is ν_e ? In no more than **two sentences**, explain how that the law of quantum mechanics requires its presence in the reaction

$$1u \rightarrow mc^2 = 931.5\text{MeV}, m_e c^2 = 511\text{keV} = 0.511\text{MeV},$$

B) Repeat part A), but for electron capture ${}^8_{14}\text{O} + \beta^- \rightarrow {}^7_{14}\text{N} + \nu_e$. Why does the relation for Q of the electron capture not include $-2m_e$?

C) Calculate the **disintegration energy** of the decay ${}^8_{16}\text{O} \rightarrow {}^Z_A\text{C} + \alpha$. Determine the value of A and Z for ${}^Z_A\text{C}$. In **one sentence**, justify, based on your answer, why the decay will or will not occur?

Question 4 Elementary Particle Physics

A) State **one (or two) reason** why the following two reactions **are not allowed**:

$$\pi^- \rightarrow e^- + \gamma \text{ and } \pi^+ \rightarrow e^- + e^+ + \mu^+$$

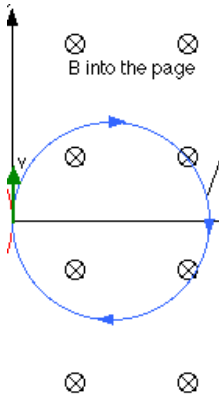
B) Complete the following reactions: $\mu^- + p \rightarrow n + ?$; $n + p \rightarrow \Sigma^0 + n + ?$. Briefly justify your answers?

C) Determine the quark composition of Λ , π^- and π^+ . See equation sheet for information on Λ .

D) Quarks are spin $\frac{1}{2}$ particles. The Ξ^0 baryon has a quark composition ssu . Is this composition consistent with the Pauli exclusion principle? Why or why not?

Question 5) Cyclotron The figure below shows a particle undergoing a **uniform circular motion** in a cyclotron of radius $r = 5.5m$ by a magnetic field of $B = 1.3T$. It was then determined that the particle has a kinetic energy of $K = 2000MeV$.

A) Using the appropriate equation, find the **relativistic momentum**, p , of the particle. **HINT:** Assume that the particle has a charge of magnitude $|q| = e$, and the small cross \otimes means that the magnetic field \vec{B} points into the page, while the charge particle follows a clockwise path.



B) Starting from $K = E - mc^2$ and $E = \sqrt{p^2c^2 + (mc^2)^2}$, show that the rest mass energy can be expressed as $mc^2 = (p^2c^2 - K^2) / 2K$.

C) Use the results of part A and B to find the rest mass energy of the particle in MeV. Inspect the nuclear physics data on the equation sheet to infer the identity of this particle. **NOTE:** Select the particle in the table 14.4 with the closest mass, and charge.