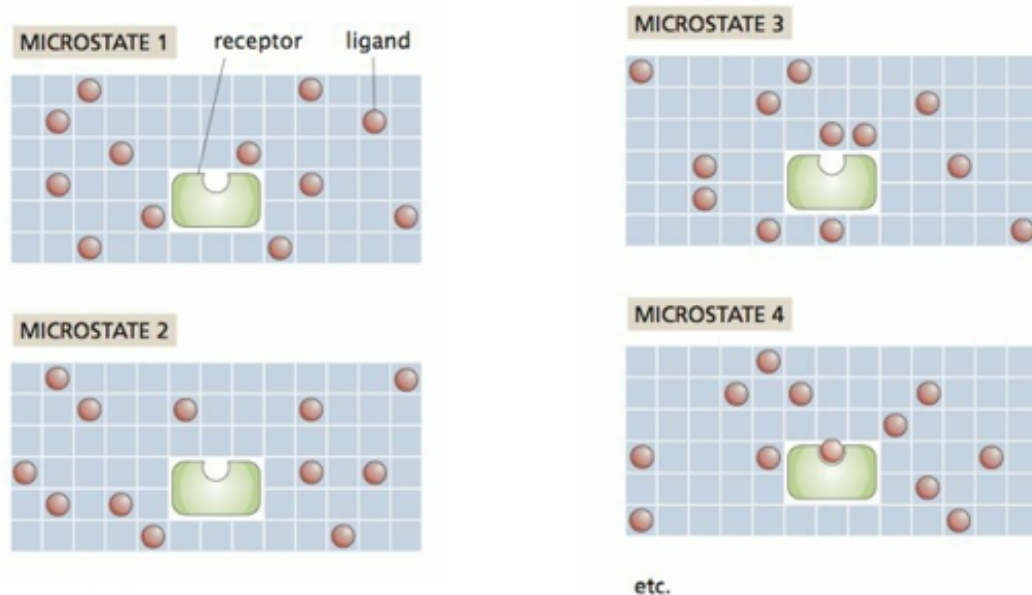


Lecture of November 5 and 7, 2018

Chapter 5 and 6: Models of Biophysical Processes

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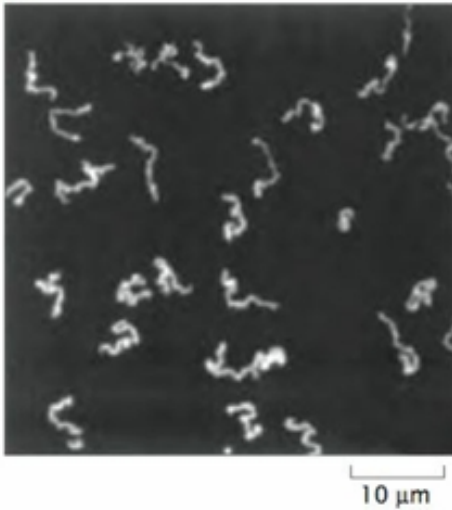
Microstates: Lattice Model of Fig 6.1



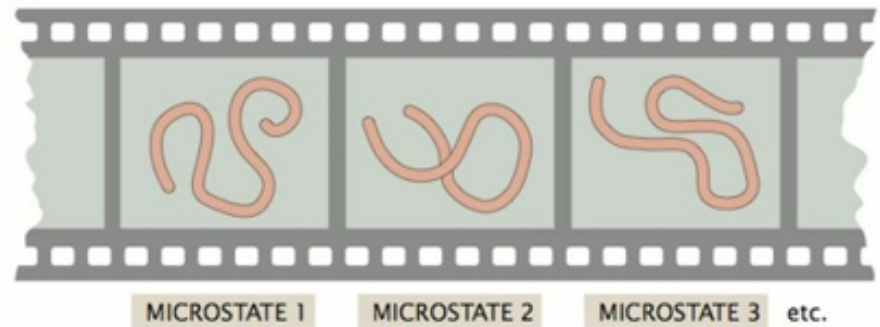
- A Microstate is a particular realization of the microscopic arrangement of the constituents for the problem of interest.
- In the above **lattice model of Ligand Receptor Binding**, the **ligands** are **spheres** that can occupy a **fix lattice positions** depicted as **squares**. There is **one receptor** (the green object) with a binding site for **one ligand**.
- In the 4 example microstates above, only microstate 4 has a bounded ligand.

Microstates of a DNA

(A)



(B)



The Canonical (constant temperature) Ensemble: Statistical Weight

For a **microstate labeled** i of energy, E_i , the occupation probability is

$$P_i = \frac{e^{-\frac{E_i}{k_B T}}}{Z}$$

with

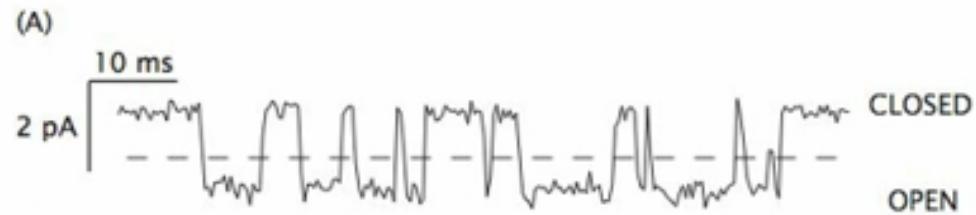
$$Z = \sum_{\text{all microstates } i} e^{-\frac{E_i}{k_B T}}$$

being the **partition function**.

$$e^{-\frac{E_i}{k_B T}}$$

is also known as the **Boltzman Factor**.

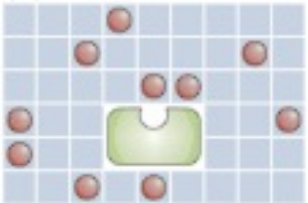
Example: Ion Channel



(B)

STATE	ENERGY	WEIGHT
	ϵ_{closed}	$e^{-\beta \epsilon_{\text{closed}}}$
	ϵ_{open}	$e^{-\beta \epsilon_{\text{open}}}$

Lattice Model of Ligand-Receptor

STATE	ENERGY	MULTIPLICITY	WEIGHT
(A) 	$L\epsilon_{sol}$	$\frac{\Omega!}{L!(\Omega-L)!} \approx \frac{\Omega^L}{L!}$	$\frac{\Omega^L}{L!} e^{-\beta L\epsilon_{sol}}$
(B) 	$(L-1)\epsilon_{sol} + \epsilon_b$	$\frac{\Omega!}{(L-1)!(\Omega-L+1)!} \approx \frac{\Omega^{L-1}}{(L-1)!}$	$\frac{\Omega^{L-1}}{(L-1)!} e^{-\beta[(L-1)\epsilon_{sol} + \epsilon_b]}$

- **Microstates A:** No bound ligand
- **Microstates B:** One bound ligand
- Note that I used **microstates** since there are many microstates with **no bound ligand**, and many with a bound ligand
- The **multiplicity** is the number of **microstates** with the **same energy**

Probability of Bound State

$$\text{weight when receptor occupied} = \underbrace{e^{-\beta \varepsilon_b}}_{\text{bound ligand}} \times \underbrace{\sum_{\text{solution}} e^{-\beta(L-1)\varepsilon_{\text{sol}}}}_{\text{free ligands}}, \quad (6.10)$$

$$p_{\text{bound}} = \frac{\sum_{\text{states}} \left(\text{grid with ligand in receptor} \right)}{\sum_{\text{states}} \left(\text{grid with ligand in receptor} \right) + \sum_{\text{states}} \left(\text{grid with ligand in solution} \right)}$$

The denominator is the partition function, Z

The partition function, Z , of the Lattice Model of Ligand-Receptor

$$Z(L, \Omega) = \underbrace{\sum_{\text{solution}} e^{-\beta L \varepsilon_{\text{sol}}}}_{\text{none bound}} + e^{-\beta \varepsilon_{\text{b}}} \underbrace{\sum_{\text{solution}} e^{-\beta (L-1) \varepsilon_{\text{sol}}}}_{\text{ligand bound}}. \quad (6.12)$$

$$Z(L, \Omega) = e^{-\beta L \varepsilon_{\text{sol}}} \frac{\Omega!}{L!(\Omega - L)!} + e^{-\beta \varepsilon_{\text{b}}} e^{-\beta (L-1) \varepsilon_{\text{sol}}} \frac{\Omega!}{(L-1)![\Omega - (L-1)]!}. \quad (6.14)$$

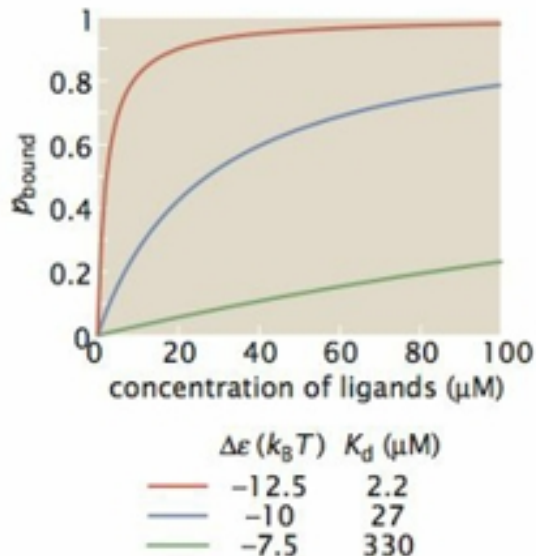
Probability of the bound state of the Lattice Model of Ligand-Receptor

$$p_{\text{bound}} = \frac{e^{-\beta\epsilon_b} \frac{\Omega^{L-1}}{(L-1)!} e^{-\beta(L-1)\epsilon_{\text{sol}}}}{\frac{\Omega^L}{L!} e^{-\beta L\epsilon_{\text{sol}}} + e^{-\beta\epsilon_b} \frac{\Omega^{L-1}}{(L-1)!} e^{-\beta(L-1)\epsilon_{\text{sol}}}}, \quad (6.17)$$

$$p_{\text{bound}} = \frac{(L/\Omega)e^{-\beta\Delta\epsilon}}{1 + (L/\Omega)e^{-\beta\Delta\epsilon}}, \quad (6.18)$$

Probability of the bound state of the Lattice Model of Ligand-Receptor

$$p_{\text{bound}} = \frac{(c/c_0)e^{-\beta\Delta\varepsilon}}{1 + (c/c_0)e^{-\beta\Delta\varepsilon}}.$$



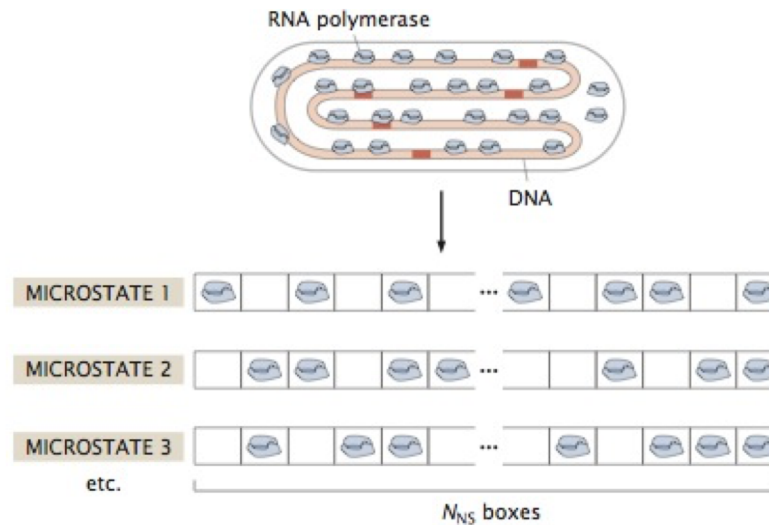
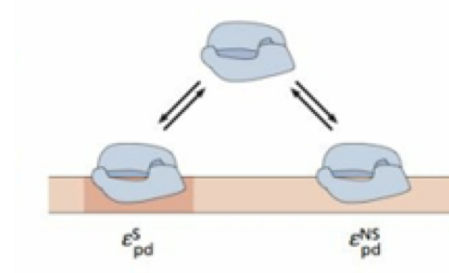
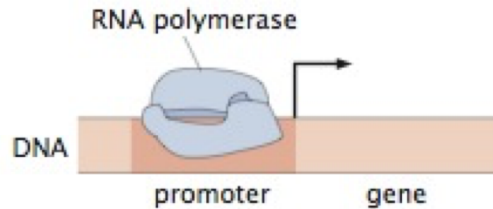
- K_D is the dissociation constant
- $c_0 = \frac{1}{V_{box}}$ is a **reference concentration**, V_{box} is the volume of **one lattice site**.
- Textbook assumes $V_{box} = 1\text{nm}^3$, which gives $c_0 \sim 0.6M$

Probability of the bound state of the Lattice Model of Ligand-Receptor

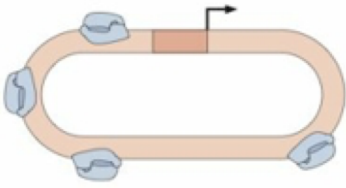
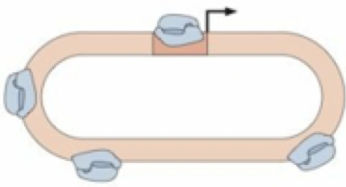
Issues:

- **Indistinguishable** Multiplicity, $\frac{\Omega!}{L!(\Omega-L)!}$
- Or **Distinguishable** Multiplicity, $\frac{\Omega!}{(\Omega-L)!}$
- Problem 6.4 and 6.5 in textbook

Binding of RNA polymerase to promoter



Binding of RNA polymerase to promoter

STATE	ENERGY	MULTIPLICITY	WEIGHT (MULTIPLICITY x BOLTZMANN WEIGHT)
	$P\epsilon_{pd}^{NS}$	$\frac{N_{NS}!}{P! (N_{NS}-P)!} = \frac{(N_{NS})^P}{P!}$	$\frac{(N_{NS})^P}{P!} e^{-P\epsilon_{pd}^{NS}/k_b T}$
	$(P-1)\epsilon_{pd}^{NS} + \epsilon_{pd}^S$	$\frac{N_{NS}!}{(P-1)! [N_{NS}-(P-1)]!} = \frac{(N_{NS})^{P-1}}{(P-1)!}$	$\frac{(N_{NS})^{P-1}}{(P-1)!} e^{-(P-1)\epsilon_{pd}^{NS}/k_b T} e^{-\epsilon_{pd}^S/k_b T}$

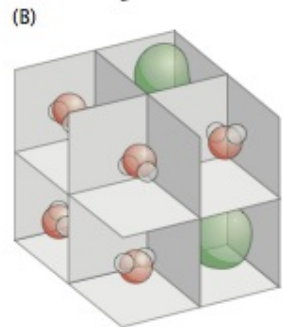
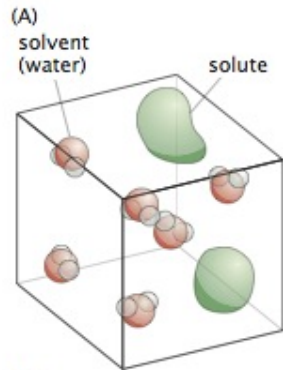
Binding of RNA polymerase to promoter

$$p_{\text{bound}} = \frac{\sum_{\text{states}} \left(\text{Diagram 1} \right)}{\sum_{\text{states}} \left(\text{Diagram 1} \right) + \sum_{\text{states}} \left(\text{Diagram 2} \right)}$$

The diagram shows a DNA double helix with RNA polymerase (represented by blue ovals) bound to a promoter region (indicated by a red box and an arrow). The DNA is shown as a series of loops, and the RNA polymerase is shown as a series of blue ovals. The promoter is labeled "promoter".

$$p_{\text{bound}} = \frac{\frac{P}{N_{\text{NS}}} e^{-\beta \Delta \epsilon_{\text{pd}}}}{1 + \frac{P}{N_{\text{NS}}} e^{-\beta \Delta \epsilon_{\text{pd}}}} = \frac{1}{1 + \frac{N_{\text{NS}}}{P} e^{\beta \Delta \epsilon_{\text{pd}}}},$$

Lattice Model of Solutes in Water H₂O



$$\mu_{\text{solute}} = \left(\frac{\partial G_{\text{tot}}}{\partial N_s} \right)_{T,p}.$$

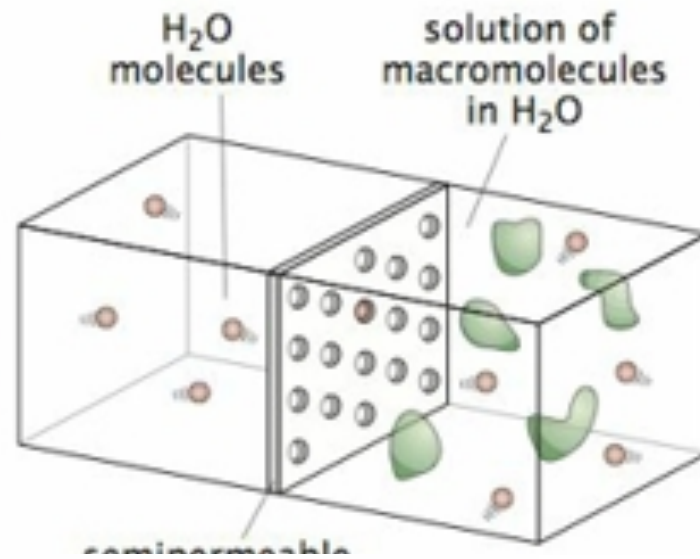
$$\mu_{\text{solute}} = G_{\text{tot}}(N_s + 1) - G_{\text{tot}}(N_s).$$

$$G_{\text{tot}} = \underbrace{N_{\text{H}_2\text{O}} \mu_{\text{H}_2\text{O}}^0}_{\text{water free energy}} + \underbrace{N_s \varepsilon_s}_{\text{solute energy}} - \underbrace{TS_{\text{mix}}}_{\text{mixing entropy}}.$$

$$S_{\text{mix}} = -k_B \left(N_{\text{H}_2\text{O}} \ln \frac{N_{\text{H}_2\text{O}}}{N_{\text{H}_2\text{O}} + N_s} + N_s \ln \frac{N_s}{N_{\text{H}_2\text{O}} + N_s} \right).$$

$$\mu_s = \varepsilon_s + k_B T \ln \frac{c}{c_0}.$$

Osmotic Pressure: Derivation of **van't Hoff's** factor



$$\underbrace{\mu_{\text{H}_2\text{O}}^0(T, p_1)}_{\text{solute-free side}} = \underbrace{\mu_{\text{H}_2\text{O}}^0(T, p_2)}_{\text{side with solutes}} - \frac{N_s}{N_{\text{H}_2\text{O}}} k_B T.$$

$$p_2 - p_1 = \frac{N_s}{V} k_B T,$$

$$\mu_{\text{H}_2\text{O}}^0(T, p_2) \approx \mu_{\text{H}_2\text{O}}^0(T, p_1) + \left(\frac{\partial \mu_{\text{H}_2\text{O}}^0}{\partial p} \right) (p_2 - p_1).$$

Van't Hoff factor for osmotic pressure, valid for dilute solution