Lecture of November 19: Chapter 7 MWC Model; Chapter 8 Polymers

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Cooperative Binding of O₂ to hemoglobin

- Earlier we discussed how hemoglobin (a protein in red blood cell, RBC) that has 4 sites for binding 4 O₂. They can also bind 4 CO₂.
- The **binding** is **cooperative**, in that binding of **oxygen** increases the **affinity** of **hemoglobin** for more **oxygen**.
- Experiments (read section 4.2, Fig 4.4 and 4.5) found that hemoglobin either bind **no oxygen** of **4 oxygen**. The binding is cooperative, or two state, in that it **all** or **nothing**.
- The molecular-level explanation is that the binding of an O₂ causes a conformational change in the hemoglobin protein.

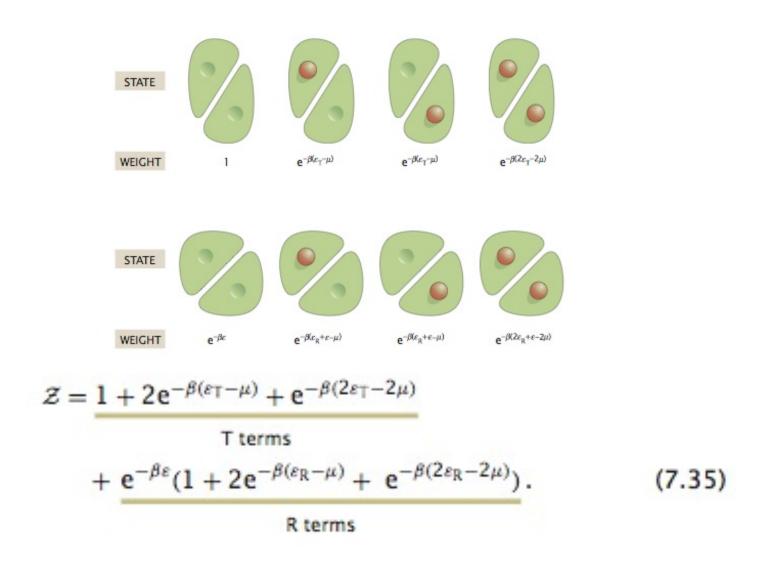
- Section 7.2.4 MWC considers a Dimoglobin that can bind **2 oxygens**.
- The dimoglobin (a protein) can be in two states: tense (T) and relaxed (R).
- In the absence of ligands the T state has lower energy than R, and T is favored.
- Variable $\sigma_m = 0.1$ for T and R, respectively

- As before $\sigma_1, \sigma_2 = 0,1$ quantifies whether a site has a bound oxygen
- Energy or Hamiltonian

$$E(\sigma_m, \sigma_i) = (1 - \sigma_m)\varepsilon_T \sum_{i=1}^2 \sigma_i + \sigma_m \left(\varepsilon + \varepsilon_R \sum_{i=1}^2 \sigma_i\right)$$

• Partition Function

$$Z = \sum_{\sigma_m = 0.1;, \sigma_i = 0, 1} exp - \beta(E(\sigma_m, \sigma_i) - N_s \mu)$$

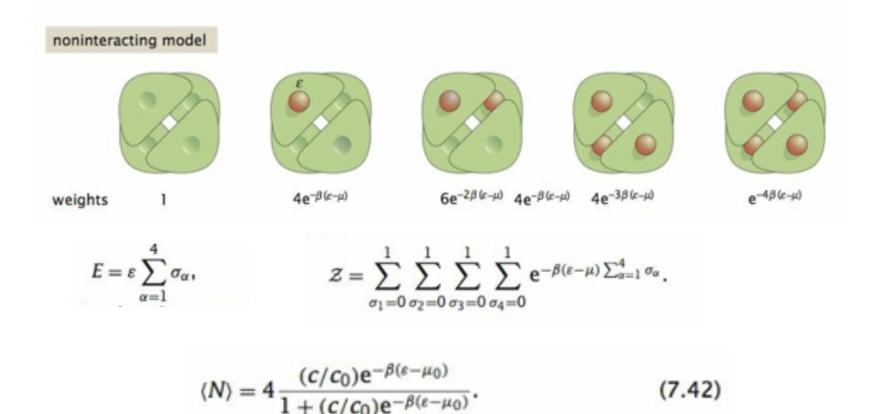


 $\mathcal{Z} = 1 + 2e^{-\beta(\varepsilon_{\mathrm{T}} - \mu)} + e^{-\beta(2\varepsilon_{\mathrm{T}} - 2\mu)}$ T terms $+ e^{-\beta\varepsilon}(1+2e^{-\beta(\varepsilon_{\mathrm{R}}-\mu)}+e^{-\beta(2\varepsilon_{\mathrm{R}}-2\mu)}).$ (7.35)R terms $\langle N \rangle = \frac{2}{\pi} [x + x^2 + e^{-\beta \varepsilon} (y + y^2)],$ $\Delta \varepsilon = -4k_0T$ 1.5 1 $\Delta \varepsilon = -2 k_{\rm R} T$ $x = (c/c_0)exp(-\beta(\varepsilon_T - \mu_0))$ $\Delta \epsilon = 0$ 0.5 $y = (c/c_0)exp(-\beta(\varepsilon_R - \mu_0))$ 0 0.5 0 $\Delta \varepsilon = \varepsilon_R - \varepsilon_L < 0$ dimensionless concentration

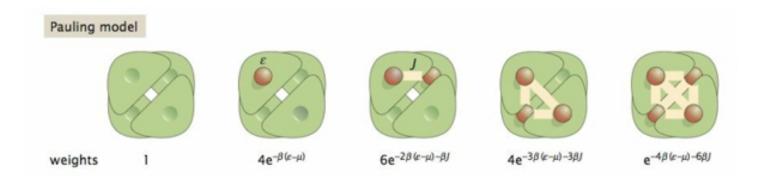
Conclusion:

- Without O_2 T state is at a higher energy of ε than the R state
- Bound Ligands make T state energetically favorable

Non-Interacting Model



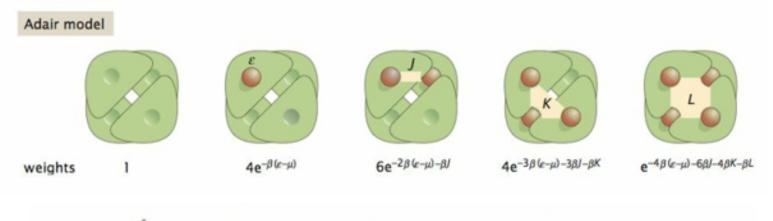
Pauling Model



$$E = \varepsilon \sum_{\alpha=1}^{4} \sigma_{\alpha} + \frac{J}{2} \sum_{(\alpha,\gamma)}' \sigma_{\alpha} \sigma_{\gamma}, \qquad \mathcal{Z} = \sum_{\sigma_{1}=0}^{1} \sum_{\sigma_{2}=0}^{1} \sum_{\sigma_{3}=0}^{1} \sum_{\sigma_{4}=0}^{1} e^{-\beta(\varepsilon-\mu) \sum_{\alpha=1}^{4} \sigma_{\alpha} - \beta(J/2) \sum_{\alpha,\gamma}' \sigma_{\alpha} \sigma_{\gamma}},$$
$$\mathcal{Z} = \underbrace{1}_{0 \text{ bound}} + \underbrace{4e^{-\beta(\varepsilon-\mu)}}_{1 \text{ bound}} + \underbrace{6e^{-2\beta(\varepsilon-\mu) - \beta J}}_{2 \text{ bound}} + \underbrace{4e^{-3\beta(\varepsilon-\mu) - 3\beta J}}_{3 \text{ bound}} + \underbrace{e^{-4\beta(\varepsilon-\mu) - 6\beta J}}_{4 \text{ bound}}.$$
$$Ae^{-\beta(\varepsilon-\mu)} + 12e^{-\beta(\varepsilon-\mu) - \beta J} + 12e^{-3\beta(\varepsilon-\mu) - 3\beta J} + Ae^{-4\beta(\varepsilon-\mu) - 6\beta J}.$$

$$\langle N \rangle = \frac{4e^{-\mu + 12e^{-\mu} + 12e^{-\mu} + 12e^{-\mu} + 12e^{-\mu} + 4e^{-3\beta(\epsilon - \mu) - 3\beta J}}{1 + 4e^{-\beta(\epsilon - \mu)} + 6e^{-2\beta(\epsilon - \mu) - \beta J} + 4e^{-3\beta(\epsilon - \mu) - 3\beta J} + e^{-4\beta(\epsilon - \mu) - 6\beta J}}.$$

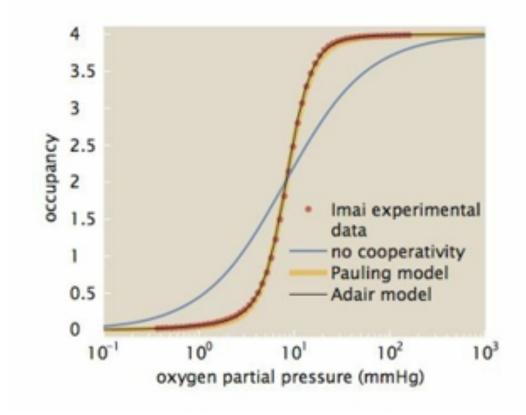
Adair Model



$$E = \varepsilon \sum_{\alpha=1}^{4} \sigma_{\alpha} + \frac{J}{2} \sum_{\alpha,\gamma}' \sigma_{\alpha} \sigma_{\gamma} + \frac{K}{3!} \sum_{\alpha,\beta,\gamma}' \sigma_{\alpha} \sigma_{\beta} \sigma_{\gamma} + \frac{L}{4!} \sum_{\alpha,\beta,\gamma,\delta}' \sigma_{\alpha} \sigma_{\beta} \sigma_{\gamma} \sigma_{\delta},$$

$$\mathcal{Z} = \underbrace{1}_{0 \text{ bound}} + \underbrace{4e^{-\beta(\varepsilon-\mu)}}_{1 \text{ bound}} + \underbrace{6e^{-2\beta(\varepsilon-\mu)-\beta J}}_{2 \text{ bound}} + \underbrace{4e^{-3\beta(\varepsilon-\mu)-3\beta J-\beta K}}_{3 \text{ bound}} + \underbrace{e^{-4\beta(\varepsilon-\mu)-6\beta J-4\beta K-\beta L}}_{4 \text{ bound}}.$$

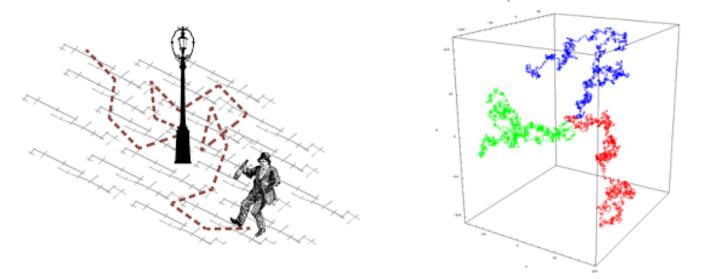
Hierarchical Hemoglobin Models



Random Walk

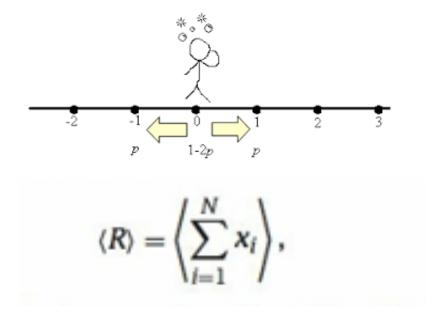
- Stochastic means having a probability distribution that may be analyzed statistically, but cannot be predicted precisely.
- A Random Walk is a Stochastic Path determined by steps whose length and direction is determined by a probability distribution.
- For large number steps the distribution is Gaussian. This is the central limit theorem.

Random Walk



- On average, after a large number of steps, the random walker position will be, $\langle x \rangle = 0$ (1D), $\langle \vec{r} \rangle = 0$ (3D)
- After a large number of steps, N, the random walker will end up far away from where he started: $\langle x^2 \rangle = NL^2$ (1D), $\langle R^2 \rangle = 2NL^2$ (3D), where L is the average step size.

1D Random Walk: Step Size a; N steps



$$\langle R^2 \rangle = Na^2$$
.

See Class Notes for Detail

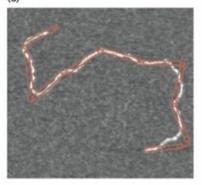
DNA, RNA and Proteins as a Random Walk

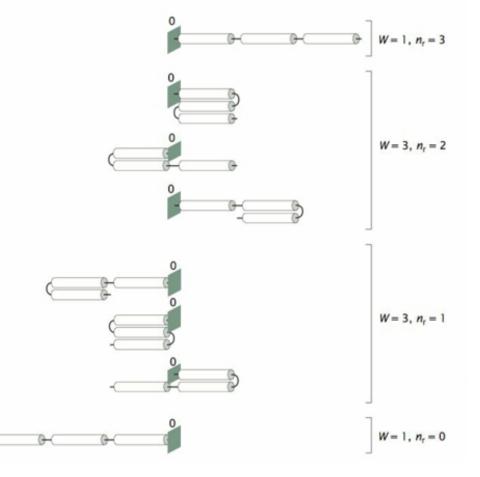
- Long molecules made of repeating units are called polymers.
- Homopolymers are polymers made up of one type of unit. An example is polyethylene (plastic)
- DNA, RNA and proteins are heteropolymers since they are made up of type of units (i.e. nucleotides, amino acids

DNA conformations as a random walk

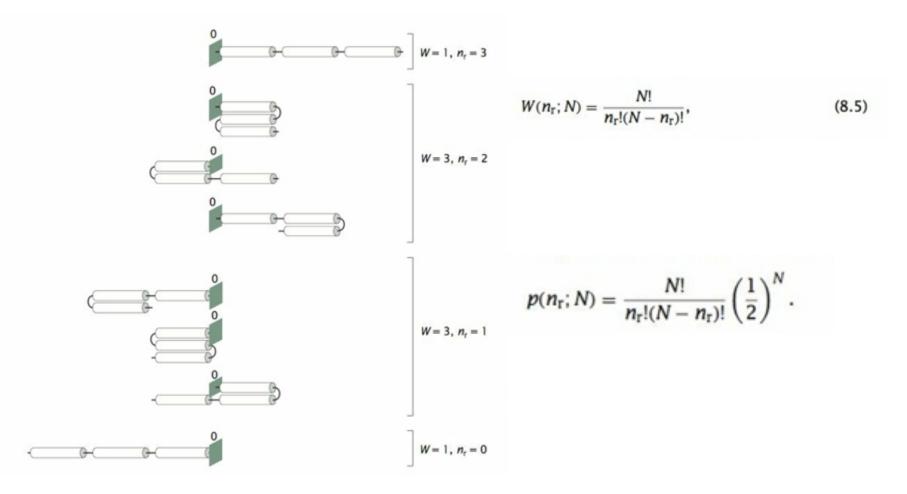
(A)

(B)





DNA conformations as a random walk



DNA end-to-end distance is a Gaussian Distribution

$$p(R; N) = \frac{2}{\sqrt{2\pi N}} e^{-R^2/2Na^2}.$$

1D Discrete probability

end-to-end distance

$$P(R; N) = \frac{1}{\sqrt{2\pi Na^2}} e^{-R^2/2Na^2},$$

1D continuous probability per unit length